Several important references were inadvertently omitted from footnote 6. This footnote should include: (f) R. E. Humphrey and J. M. Hawkins, *Anal. Chem.*, **36,** 1812 (1964); (g) R. E. Humphrey, A. L. McCrary, and R. M. Webb, *Talanta,* **12,** 727 (1965); (h) R. E. Humphrey and J. L. Potter, *Anal. Chem.*, **37,** 164 (1965); (i) B. J. Sweetman and J. A. Maclaren, *Aust. J. Chem.*, **19,** 2347 (1966); (j) M. Grayson and C. E. Farley, *Chem. Commun.*, 831 (1967); (k) M. Grayson, *Colloq. Int. C.N.R.S.*, **182,** 275 (1969).

Reactions of π-2-Methallylnickel Bromide with Methylbenzoquinones. Evidence for Electron Transfer [*J. Am. Chem. Soc.*, **96,** 6789 (1974)]. By L. S. HEGEDUS* and E. I. WATERMAN, Department of Chemistry, Colorado State University, Fort Collins, Colorado 80521.

On page 6790, two lines were transposed in the paragraph beginning at the bottom of the left-hand column. The paragraph should read:

Finally, the *site* of alkylation of the various methylbenzoquinones studied corresponds to the *noncarbonyl ring site of highest spin density* in the corresponding quinone radical anion, as measured from esr hyperfine splitting constants and as calculated by Hückel LCAO-MO methods.[5-7] Table I shows this correlation by listing the calculated spin densities for the various quinone radical anions followed by the allyl products obtained from the reactions summarized in eq 1.[8] . . .

A New Ring System. 2,6-Dioxabicyclo[2.2.2]octane, a Highly Reactive Bicyclic Acetal [*J. Am. Chem. Soc.*, **96,** 7265 (1974)]. By H. K. HALL, JR.,* L. J. CARR, R. KELLMAN, and F. DE BLAUWE, Department of Chemistry, University of Arizona, Tucson, Arizona 85721.

"A New Ring System" should be deleted from the title.

In the first sentence of the Abstract, change "of a new ring system" to "of a strained ring system".

In the first sentence of the Discussion (p 7266), change "this ring system" to "this compound".

Diol Dehydrase Model Studies. The Acid Catalyzed Rearrangement of β-Hydroxyisopropylcobaloxime [*J. Am. Chem. Soc.*, **96,** 7681 (1974)]. By KENNETH L. BROWN and LLOYD L. INGRAHAM,* Department of Biochemistry and Biophysics, The University of California, Davis, California 95616.

On page 7684 we state that Schrauzer and Windgassen probably did not obtain pure β-hydroxyisopropylcobaloxime in basic solution from a 2-halo-1-propanol. We have subsequently been informed by Dr. Schrauzer that he did obtain the pure product. The necessary structural proof was obtained but not published with the synthesis. Accordingly, we retract our statement.

Hydroboration. XXXVI. A Direct Route to 9-Borabicyclo[3.3.1]nonane *via* the Cyclic Hydroboration of 1,5-Cyclooctadiene. 9-Borabicyclo[3.3.1]nonane as a Uniquely Selective Reagent for the Hydroboration of Olefins [*J. Am. Chem. Soc.*, **96,** 7765 (1974)]. By HERBERT C. BROWN,* EVORD F. KNIGHTS, and CHARLES G. SCOUTEN, The Richard B. Wetherill Laboratory of Purdue University, West Lafayette, Indiana 47907.

In Table III the products from styrene are 2-phenylethanol, 98.5%; 1-phenylethanol, 1.5%. (The printed prefixes are transposed.)

Synthesis, Characterization, and Bonding of Tetrameric Triphenylphosphine Silver Halide Cluster Systems. Evidence of Dictation of Stereochemistries by van der Waals Interactions [*J. Am. Chem. Soc.*, **97,** 1256 (1975)]. By BOON-KENG TEO,* Bell Laboratories, Murray Hill, New Jersey 07974, and JOSEPH C. CALABRESE, Department of Chemistry, University of Wisconsin, Madison, Wisconsin 53706.

Between the bottom of page 1256 and the top of page 1257, two lines have been omitted. This passage should read "the average Ag–X distances are substantially longer than normal covalent bonds by 0.32 and 0.24 Å in **1** and **2a,** respectively. The observed lengthening of the Ag . . . Ag distances . . . ."

Proton-Exchange Reactions of Acetone and Butanone. Resolution of Steps in Catalysis by Acetoacetate Decarboxylase [*J. Am. Chem. Soc.*, **97,** 1568 (1975)]. By GEORGE HAMMONS, F. H. WESTHEIMER,* KURT NAKAOKA, and RONALD KLUGER,* James Bryant Conant Laboratories, Harvard University, Cambridge, Massachusetts 02138, and the Department of Chemistry, University of Chicago, Chicago, Illinois 60637.

The caption to Figure 3 should read: (a) The 60-MHz spectrum of the protons of the 4 position of butanone after ~55% of the original signal at the 3 position remains, catalyzed by hydroxide; (b) the same except catalyzed by enzyme.